Microbial Community Dynamics during the Biochemical Treatment of Acid Mine Drainage under three different Hydraulic Retention Times

Yaneth Vasquez¹, María C. Escobar¹, Carmen M. Neculita², Ziv Arbeli³, Fabio Roldan³

¹Departamento de Ciencias Naturales, Universidad Central, Bogotá, Colombia, ovasquezo@ucentral.edu.co ² RIME - Research Institute on Mines and Environment, UQAT - University of Quebec in Abitibi-Temiscamingue, QC, Canada, Carmen-Mihaela.Neculita@uqat.ca ³Unidad de Saneamiento y Biotecnología Ambiental (USBA), Departamento de Biología, Pontificia Universidad Javeriana, Bogotá, Colombia, fabio.roldan@javeriana.edu.co

Abstract

In the Zipaquirá Mining District of Colombia, there are about 600 coal mines that generate 70,400 m³/month of acid mine drainage. A sustainable approach to remediate AMD is to use biochemical passive reactors. However, limited data is available on the dynamics (temporal and spatial) of microbial community and their activity under different hydraulic retention times, in the long-term operation of BPR. Seven 5L biochemical passive reactors (73×10 cm) were operated during 36 weeks, under three different hydraulic retention time (1, 2, and 4 days). The reactors were sacrificed on 8, 17 and 36 weeks, and the reactive mixture was sampled at the bottom, middle, and top layers. The microbial community of the post-treatment reactive mixtures was monitored by sequencing (Illumina MiSeq) and correlated with physicochemical parameters. The result showed that operation time, location and hydraulic retention time had significant effects on physicochemical changes of the reactive mixture and it is rather the combination of factors affect diversity during the AMD treatment. In addition, the microbial community analysis resulted in the identification of specialized groups related to cellulose degraders and fermentative bacteria that work in synergy for degrading substrate make the organic material available to sulfate-reducing bacteria. This microbial community analysis provides a base line for future studies in the BPR

Key words: acid mine drainage; microbial diversity; Illumina; biochemical passive reactors

Introduction

Biochemical passive reactors (BPR) is a successful acid mine drainage (AMD) treatment technology with potential advantages such as low costs, few site visits required, ability to work in remote areas, opportunities to use recycled or waste materials, and natural appearance (Doshi 2006). In the Zipaquirá Mining District of Colombia, there are about 600 coal mines that generate ~70,400 m³/month of drainages and in this region a sustainable approach to remediate AMD is to use BPR. The most efficient reactive mixture for increasing pH and alkalinity, as well as promoting sulfate reduction and metal removal during AMD treatment in Zipaquirá Mining District was selected (Vasquez et al 2016a). In addition, the effect (temporal and spatial) of hydraulic retention time (HRT) (1, 2 and 4 days) on the efficiency of BPRs and microbial activity was also evaluated (Vasquez et al 2016b). However, the microbial community dynamics of this system has not been characterized despite its importance for BPR.

The HRT is a crucial design parameter, which influences the overall performance of BPR during AMD treatment (Neculita et al 2008a). Nevertheless, little is known about how the HRT affects microbial communities during operation time in BPR. In this context, the objective of the present study was to assess the impacts of HRT, location in the reactor and operation time on the microbial community involved in the synthetic AMD remediation under a column study.

Materials and methods

BPR design and AMD characteristics

Seven up-flows BPR were constructed using acrylic columns (73×10 cm) and operated for 36 weeks treating synthetic AMD, characterized by high sulfate concentrations, and low metal loading (mg L⁻¹ 201 ± 44 Fe²⁺; 30 ± 2 Mn²⁺; 19 ± 2 Zn²⁺; 215 ± 11 Ca²⁺; 128 ± 13 Mg²⁺ and 2,500 ± 105 SO₄²⁻, at pH 3.0 – 3.7). The AMD was prepared according the information collected at five active mine sites in the Zipaquirá Mining District, Colombia. The columns were filled with the same reactive mixture (15% cow manure, 10% mushroom compost, 25% sajo sawdust, 15% gravel, 20% limestone, and 15% wetland sediment as inoculum). Initially three BPR were operated with 2-days of HRT and four BPR with 4-days of HRT. After 17 weeks, a strong increase of soluble sulfide in treated effluents from the columns with 4-day HRT justified the decision to change one of the reactors of 4-day HRT to 1-day HRT (Vasquez et al 2016 a, b).

Column sampling

Columns were sacrificed throughout the study to monitor the changes in the post-treatment reactive mixture and microbial activity. Four columns, two of 2-day HRT and two of 4-day HRT, were sacrificed at week 8 and 17. The other three columns (1, 2, and 4-day HRT) were analyzed at the end of the study (36 weeks). The reactive mixture from the sacrificed columns was removed from different locations at the reactor. Three layers (20×10 cm), one from the bottom (at 0–20 cm), one from the middle (at 20–40 cm), and one from the top (at 40–60 cm) were taken. The samples were homogenized and refrigerated at 4 °C for physicochemical analyses and stored at – 80°C until DNA extraction.

Physicochemical analysis of reactive mixture post treatment

The pH was measured (Lab 870, Schott; Mainz, Germany) using method 4972–01 (ASTM 1995a) and a ratio of solid to de-ionized water of 1:1. The organic nitrogen (TKN) was measured by Method 4500–Norg (APHA 2005), and cellulose content was determined, according to Harper and Lynch (1981). The organic carbon (TOC) was analyzed by the Walkley–Black method (Schumacher 2002). Total metal concentrations immobilized in the reactive mixtures were determined by the digestion method described by Neculita et al (2008b) and dissolved metals (Ca²⁺, Mg²⁺, Fe²⁺, Mn²⁺ and Zn²⁺) were quantified by atomic absorption spectrometry (Varian 240 FS; Agilent Technologies; Santa Clara, CA), using method 7000B (USEPA 2007). Acid volatile sulfide (AVS) were separated by Brouwer & Murphy (1994) and soluble sulfate were extracted by Sobek et al (1978) and quantified by UV-VIS spectrophotometry (Genesys 10, Thermo Scientific; Waltham, MA) using method 4500–SO4 (APHA 2005).

Nucleic acid extraction and sequence analysis

Genomic DNA was extracted from 22 samples of post-treatment reactive mixture and 1 sample of initial reactive mixture using the MoBio® PowerSoil DNA extraction kit (MoBio Laboratories, Solana Beach, CA). PCR amplification, purification, and sequencing for illumina MiSeq of a region V4 of the 16S rRNA gene were performed following the procedure described by Caporaso et al (2011). All extractions and amplifications were realized by triplicated. Sequencing was conducted using MiSeq Illumina (2×250 pb) technology at DNA Facilities (Iowa University). Total length of the Paired-end reads (250 bp) were assembled with the Fast Length Adjustment of Short Reads tool (Magoc and Salzberg 2011). QIIME v1.7 was used for all analysis and the sequences were aligned to the Greengenes reference alignment using PyNAST at the 97% confidence level. After the sequences were quality filtered and randomly in subsampled the 10000 sequences (this number was chosen by the minimum number on reads in a control sample) that were subsequently clustered into operational taxonomic units (OTUs). The relationships between dynamic of genera (relative abundance > 0.5%), physicochemical characteristics of BPR and samples were assessed by canonical correspondence analysis (CCA) using Conoco v4.5 for Windows package with Monte Carlo permutation test and Spearman correlation coefficients. The statistic difference of the relative abundance of the genera through the layers of the BPR was evaluated by t -test Welch's, adjusted for Benjamini Hochberg (p = 0.05), using STAMP 2.01 (Parcks et al 2010).

Results and discussion

The CCA (Fig. 1) reveled that samples were clustered in three subgroups corresponding to the operation time (8, 17 and 36 weeks) and that these subgroups were significantly different with respect to their physicochemical characteristics and microbial community. Besides, the Spearman correlation coefficients showed that operation time had negative correlation with pH, TKN and COT (-0.884, -0.845 and -0.812; p =0.00, respectively) and positive correlation with Zn, AVS, sulfate, Fe and Ca (0.817, 0.771, 0.742, 0.455, and 0.406; p < 0.05, respectively), indicating that organic components and pH decrease while the metal sulfides increase in reactive mixture over time. Operation time is one driver for shifting the physicochemical characteristics and microbial community composition. Previous studies have identified that lowering of the pH, re-oxidation of sulfide back to sulfate and a reduction in dissolved organic compounds contributed to change in the microbial community and decline in performance of the BPR (Baldwin et al 2015; Mirjafari et al 2011).

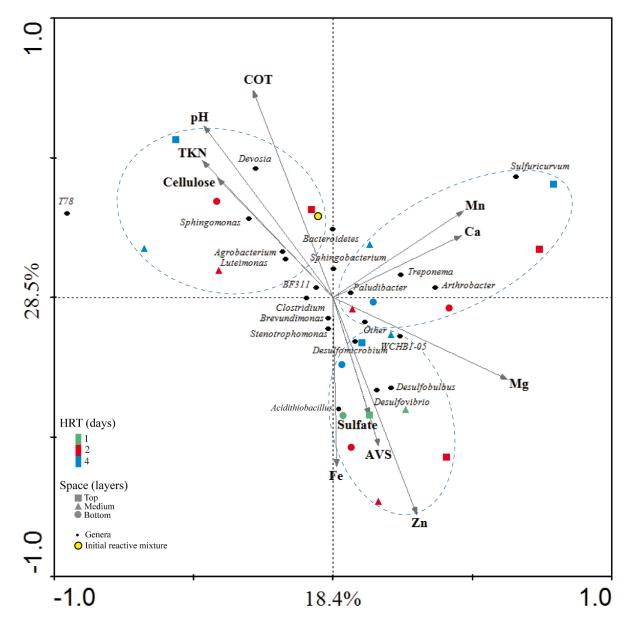


Figure 1 Relation between the relative abundance of genera (>0.5%), physicochemical parameters and samples. CCA-triplot where the x-axis explains 18.4% and the y-axis explain 28.5% of the variation. The colors indicate samples with different HRT (1, 2 and 4 days) and its form (square, triangle and circle) show the location in BPR.

The first group that emerge from the CCA correspond to the initial reactive mixture and 8 week samples. These samples were positively correlated (Spearman, p < 0.05) with pH (> 8.0) and as TKN, COT and cellulose (> 6.0, > 33.0 and > 50.7 % w/w, respectively) without difference significate between layers in the BPR. This group included the genera *Devosia* (4.6%), *Sphingobacterium* (4.2%), *Sphingomonas* (4.1%), *Agrobacterium* (3.8%), *Luteimonas* (2.0%) and *Bacteroides* (1.3 %). Members of these genera have been studied for their ability to degrade plant cell wall material and they utilize a wide variety of compounds as carbon and energy sources, including cellulose, hemicellulose, starch, and pectin. These genera have been reported in previous studies and its presence in the reactive mixture is considered an advantage during AMD treatment (Drennan et al 2015; Hiibel et al 2011). Other genera with low relative abundance (< 1.0%) were *Treponema*, *clostridium*, *Desulfovibrio*, *Desulfomicrobium* and *Desulfobacter*. The presence of SRB in the initial reactive mixture shortened the initial lag phase of the AMD treatment and contributes with low cost and improve the performance of BPR (Mirjafari et al, 2014).

Other group clustered in CCA correspond to samples extracted during the second sacrifice (17 week). In this group the genus more abundant was *Treponema* (10.8%) with significant abundance in medium layer for 4-day of HRT (t -test Welch's < 0.01). This genus has been observed to perform acetogenesis, carbon fixation and it is often associated with cellulose degradation (Do et al 2014; Sanchez-Andrea et al 2014). The second genus with the high abundance was Sulfuricurvum (6.4%, on average) with significant abundance in the top layer of columns with 2-day of HRT. These sulfur-oxidizing bacteria produces adverse effects in BPR because can oxidize the sulfide to sulfate (Zheng et al, 2014). Its presence possibility was due to change of pipeline in sampling ports located at the top cap of biorreactor allowing the formation of microaerobic regions. Other genera which also increased their abundance were *Desulfovibrio* (1.5%) and *Desulfomicrobium* (1.1%) with significant difference in BPR for 4-day of HRT. This increased of BSR favored that the concentration of sulfides $(2,826 \pm 185 \text{ mg H}_2\text{S L}^1)$ was higher in this BPR because longer residence time allowed greater oxidation of available organic carbon and reduction of sulfate. In the week 17, the concentration of Ca (46 mg kg⁻¹) and Mn (< 1.0 mg kg⁻¹) in reactive mixture post treatment were the physicochemical parameter with higher effect on microbial community (0.406 and 0.502; p < 0.05, respectively). The effect of Ca could be related with the loss of nutrients in solution for formation of colloidal suspensions which precipitate making difficult the access for microorganism (Lindsay et al 2011). On the other hand, during 17 week the Mn presented low concentration in the reactive mixture post treatment and high concentration in the effluents (77 ± 4 mg L^{-1} for 4-day HRT and 60 ± 2 mg L^{-1} for 2-day HRT) which exceeded levels in the synthetic AMD (31) mg L^{-1}). This metal only was removed of the AMD at the beginning of the treatment when it was probably adsorbed on the reactive mixture but after it released causing toxicity on microbial community.

In the week 36, three bioreactors with different HRT (1, 2 and 4 d), were sacrificed. The genera most abundance in the columns with 2 and 4-day of HRT were Treponema (8.2% on average), and Paludibacter (3.2% on average). Previous studies demonstrated that Paludibacter, as fermentative bacteria, had appeared in enrichments of sulfate reduction systems in acidic condition (Zheng et al, 2014; Sánchez-Andrea et al, 2014). The Paludibacter was often accompanied by the SRB in sulfate reduction systems (Lindsay et al 2011). The presence of acid lactic producing bacteria genera as Treponema and Paludibacter suggests a potential for suitable SRB electron donor production in the system (Dennan et al 20165). In the column with 1-day of HRT, Acidithiobacillus (10.5% on average) was the genus with the most abundance and its presence was related with decline in pH (5.2), in the bottom layer of the reactive mixture. Members of these genera are frequently found in metal-rich acidic environments associated with metal sulfide leaching (Garcia-Moyano et al 2008). Acidophilic chemolithotrophic microorganisms play a key role maintaining a high concentration of ferric iron in AMD (Sánchez-Andrea et al 2014). In the week 36, the genera of SRB increased in the three bioreactors with significate difference in 4-day of HRT, with Desulfovibrio (1.5%) as most abundance follow by Desulfomicrobium (0.7%), Desulfobulbus (0.6%), and Syntrophobacter (0.5%). The CCA analysis showed correlation (Monte Carlo, p < 0.05) between sulfide, Fe, AVS and the genera *Desulfococcus*, Desulfobulbus, Desulfomona, Desulfobacter and Desulfovibrio. This genera of BSR were most abundance in bottom layer of BPR and its presence has been reported in rich environments with metal sulfides (Hao *et al.*, 2014). Besides, this BSR have been found in BPR with low pH (< 5.0) and sediments from acid sites (Sánchez-Andrea et al 2014).

Conclusions

The findings in the present study provides critical information regarding dynamic of the microbial community present in BPR during the treatment of AMD. The operation time (8, 17 and 36 weeks), the space (top, medium and bottom) and the HRT (1, 2 and 4-day) had significant effects on physicochemical changes of the reactive mixture of BPR and these changes affected the diversity and the abundance relative (> 0.5%) of microbial community during AMD remediation. In addition, the microbial community analysis resulted in the identification of specialized groups related to cellulose degraders and fermentative bacteria that work in synergy for degrading substrate make the organic material available to sulfate-reducing bacteria. Finally, the microorganisms associated with metal-rich waters were identified with roles in the iron and sulfur cycles of AMD communities. This analysis provides a base line for future studies in field BPR.

Acknowledgements

This research was funded by the Pontificia Universidad Javeriana (ID. 5177) and the Department of Natural Sciences of the Universidad Central. We thank Johan Saenz for assistance with bioinformatics analysis.

References

- APHA (2005) Standard methods for the examination of water and wastewater (21st Ed.). L.S. Clesceri, A.E. Greenberg, A.D. Eaton (Eds.), American Public Health Association Washington, D.C.
- ASTM (1995a) Standard test method for pH of soils. In: Annual book of ASTM standards, Section D4972-95a, ASTM International, West Conshohocken, PA, pp. 27–28.
- Baldwin SA, Khoshnoodi M, Rezadehbashi M, Taupp M, Hallam S, Mattes A, Sanei H (2015) The Microbial Community of a Passive Biochemical Reactor Treating Arsenic, Zinc, and Sulfate-Rich Seepage. Frontiers in Bioengineering and Biotechnology, 3
- Brouwer H, & Murphy T P. (1994) Diffusion method for the determination of acid-volatile sulfides (AVS) in sediment. Environmental Toxicology and Chemistry, 13(8), 1273-1275.
- Caporaso J, Lauber C, Walters W, Berg-Lyons D, Lozupone C, Turnbaugh P, Fierer N, Knight R (2011) Global patterns of 16S rRNA diversity at a depth of millions of sequences per sample. Proceeding of the National Academy of Sciences of the United States of America. U. S. A., 108:4516–22.
- Do TH, Nguyen TT, Nguyen TN, Le QG, Nguyen C, Kimura K, Truong NH (2014) Mining biomass-degrading genes through Illumina-based de novo sequencing and metagenomic analysis of free-living bacteria in the gut of the lower termite Coptotermes gestroi harvested in Vietnam. *Journal of bioscience and bioengineering*, *118*(6), 665-671.
- Doshi SM (2006) Bioremediation of acid mine drainage using sulfate- reducing bacteria. Prepared by National Network of Environmental Management Studies for US EPA, Washington, DC
- Drennan DM, Almstrand R, Lee I, Landkamer L, Figueroa L, Sharp JO (2015) Organoheterotrophic bacterial abundance associates with zinc removal in lignocellulose-based sulfate-reducing systems. Environmental science & technology, 50(1), 378-387.
- García-Moyano A, Gonzalez-Toril E, Aguilera Á, Amils R (2012) Comparative microbial ecology study of the sediments and the water column of the Río Tinto, an extreme acidic environment. FEMS Microbiology Ecology, 81(2), 303-314.
- Harper S, Lynch J (1981) The chemical components and decomposition of wheat straw leaves, internodes and nodes. Journal of the Science of Food and Agriculture, 32, 1057–1062.
- Hiibel SR, Pereyra LP, Breazeal MV, Reisman DJ, Reardon KF, Pruden A (2011) Effect of Organic Substrate on the Microbial Community Structure in Pilot-Scale Sulfate-Reducing Biochemical Reactors Treating Mine Drainage. Environmental Engineering Science, 28(8), 563–572.
- Lindsay M, Wakeman K, Rowe F, Grail B, Ptacek C, Blowes D, Johnson D (2011) Microbiology and geochemistry of mine tailings amended with organic carbon for passive treatment of pore water. Geomicrobiology Journal .28 229-241.

- Magoc T, Salzberg SL (2011). FLASH: Fast Length Adjustment of Short Reads to Improve Genome Assemblies. Bioinformatics, 1–8.
- Mirjafari P, Rezadehbashi M, Baldwin SA (2012) Microbial Communities Associated with Passive Treatment of Sulphate and Selenium Containing Water. In International Conference on Acid Mine Drainage, May 20-26, Ottawa, Canada.
- Neculita CM, Zagury GJ, Bussière B (2008a) Effectiveness of sulfate-reducing passive bioreactors for treating highly contaminated acid mine drainage: I. Effect of hydraulic retention time. Applied Geochemistry, 23(12), 3442-3451.
- Neculita CM, Zagury GJ, Bussière B (2008b) Effectiveness of sulfate-reducing passive bioreactors for treating highly contaminated acid mine drainage: II. Metal removal mechanisms and potential mobility. Applied Geochemistry, 23(12), 3545-3560.
- Park JH, Lamb D, Paneerselvam P, Choppala G, Bolan N, Chung JW (2011) Role of organic amendments on enhanced bioremediation of heavy metal (loid) contaminated soils. Journal of Hazardous Materials, 185, 549–574.
- Sánchez-Andrea I, Sanz JL, Bijmans M, Stams A (2014) Sulfate reduction at low pH to remediate acid mine drainage. Journal of Hazardous Materials, 269(3), 98–109.
- Schumacher BA, (2002) Methods for the determination of total organic carbon (TOC) in soils and sediments. NCEA-C-1282, EMASC-001. Ecological Risk Assessment Support Center, Office of Research and Development, U.S. Environmental Protection Agency, Las Vegas, NV.
- Sobek AA, Schuller WA, Freeman JR (1978) Field and laboratory methods applicable to overburdens and minesoils. In Field and laboratory methods applicable to overburdens and minesoils. EPA.
- United States Environmental Protection Agency (USEPA) (2001) Flame atomic absorption spectrophotometry. Method 7000b, US Environmental Protection Agency, Washington, D.C.
- Vasquez Y, Escobar MC, Neculita CM, Arbeli Z, Roldan F (2016) Selection of reactive mixture for biochemical passive treatment of acid mine drainage. Environmental Earth Sciences, 75(7), 1-11.
- Vasquez Y, Escobar MC, Neculita CM, Arbeli Z, Roldan F (2016) Biochemical passive reactors for treatment of acid mine drainage: Effect of hydraulic retention time on changes in efficiency, composition of reactive mixture, and microbial activity. Chemosphere, 153, 244-253.
- Zheng Y, Xiao Y, Yang ZH, Wu S, Xu HJ, Liang FY, Zhao F (2014) The bacterial communities of bioelectrochemical systems associated with the sulfate removal under different pHs. Process Biochemistry, 49(8), 1345-1351.